



SE-6883

B. E. III (Sem. V) (TP) Examination

April / May - 2011

Analytical Textile Chemistry - I

(Old Scheme)

Time : 3 Hours]

[Total Marks : 100

Instructions :

(1)

नीचे दृष्टावेक निशानीवाणी विगतो उत्तरवडी पर अवश्य बपवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="text" value="B. E. 3 (Sem - 5) (TP)"/>	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="text" value="Analytical Textile Chemistry - 1"/>	<input type="text"/>
Subject Code No. : <input type="text" value="6"/> <input type="text" value="8"/> <input type="text" value="8"/> <input type="text" value="3"/>	<input type="text"/>
Section No. (1, 2,.....) : <input type="text" value="Nil"/>	<input type="text"/>
	Student's Signature

(2) Figures to the right indicate full marks.

- 1 (a) Answer the following objective questions : 10
- (1) Colourimetric determinations are carried out at λ max. true or false ?
 - (2) KCNS is used to determine _____ content of water.
 - (3) Colour of water is measured in ppm, true or false ?
 - (4) Name two sequestering agents.
 - (5) Soft waters generally have neutral pH, true or false ?
 - (6) Ash content indicates _____
 - (7) Iodine value indicates _____ content of oil.
 - (8) Nitrogen content is measured by _____ method.
 - (9) Spectrophotometry follows _____ and _____ laws.
 - (10) What does the acid number indicate ?
- (b) What is meant by hard water ? Classify hardness in water. Give the method for determination of total of hardness water. 10
- 2 Explain the terms COD and BOD. Give the method for determination of D.O. and BOD of an effluent. 15

OR

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[Contd...

- 2 Discuss, conductometric and potentiometric titrations and their applications in analytical textile field. 15
- 3 Write short notes : (any three) 15
- (a) Silica content determination
 - (b) Iron content and its effect in processing
 - (c) % purity of H_2O_2 and NaOCl
 - (d) Brooke field viscometer
- 4 (a) Answer the following objective questions : 10
- (1) Rain water is natural distilled water. true or false ?
 - (2) BOD is always greater then COD. true or false ?
 - (3) BOD 3 means ?
 - (4) _____ reagent used for chloride content in water.
 - (5) Turbidity of water is measured in terms of _____ .
 - (6) "Boiling removes total hardness of water" Correct the sentence.
 - (7) Flash point and fire point can be measured by _____ cup and _____ cup method.
 - (8) What is meant by bomb wastings ?
 - (9) A grease sample must be evaluated for _____ number.
 - (10) What is meant by standard electrode petential ?
- (b) Describe working of Brookfield viscometer and Redwood viscometer. 10
- 5 Describe the methods for testing of lubricants for consistency, aniline point, cloud point, power point, flash point and fire point. 15

OR

- 5 Explain the method for determination of calorific value of coal using bomb-calorimeter with neat sketch and calculations. 15
- 6 Write short notes : (any three) 15
- (a) TLC
 - (b) Principle of spectroscopy and its applications
 - (c) Solids in water and their determination
 - (d) Determination of COD.